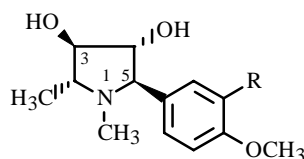


## STEREOCHEMISTRY OF THE ALKALOIDS CODONOPSINE AND CODONOPSININE

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The isolation of the alkaloids codonopsine (**1**) and codonopsinine (**2**) from the aerial part of *Codonopsis clematidea* (harebell, Campanulaceae) has been previously reported [1]. Based on spectral data (IR, UV, NMR, and mass spectroscopy) and chemical transformations (Hofmann degradation,  $\text{KMnO}_4$  and periodic-acid oxidation), the structures for **1** and **2** were established as 1,2-dimethyl-3,4-dihydroxy-5-(3,4-dimethoxyphenyl)pyrrolidine and 1,2-dimethyl-3,4-dihydroxy-5-(4-methoxyphenyl)pyrrolidine, respectively [2, 3]. Later double-resonance PMR spectroscopy was used in an attempt to determine the relative configuration of the substituents on the N-methylpyrrolidine ring [4].



**1, 2**

**1:** R = OCH<sub>3</sub>

**2:** R = H

The methyl iodide of codonopsine (**3**) was prepared and its x-ray crystal structure was solved in order to solve unambiguously these issues.

The results of the x-ray structure analysis (XSA) confirm the proposed chemical formulas for **1** and **2** [2, 3] but necessitate a correction in the stereochemistry of the N-methylpyrrolidine ring. Figure 1 shows the corrected structural formulas for **1** and **2** and the molecular structure of codonopsine methyl iodide (**3**).

It can be seen that the C2 methyl and the C4 hydroxyl have the  $\alpha$ -configuration; the C3 hydroxyl and the C5 substituted aromatic ring, the  $\beta$ -configuration. The N-methylpyrrolidine ring adopts the  $5\alpha$ -envelope conformation. Atom C5 deviates from the plane of the other four atoms (coplanar within  $\pm 0.027 \text{ \AA}$ ) by  $0.625 \text{ \AA}$ . For this reason, only C4 and C5 can have substituents that are pseudo-axial or -equatorial relative to the plane of the four atoms. The C4 hydroxyl and C5 aromatic ring are pseudo-equatorial. The planar ( $\pm 0.01 \text{ \AA}$ ) substituted aromatic ring is twisted relative to the plane of the pyrrolidine ring by  $74.2^\circ$ .

The absolute configuration of **3** was also established from the XSA. According to the Flack parameter [5] [0.00(4)], the absolute configuration of the chiral centers corresponds with that shown in Fig. 1. Therefore, all asymmetric centers (C2, C3, C4, and C5) have the *R*-configuration.

The crystal packing of **3** indicates that I...H-O and O...H-O H-bonds are present. The parameters for the I...H-O1 bond are: I...O1  $3.45$ , I...H-O1  $2.66 \text{ \AA}$  and I...H-O1  $172.1^\circ$ . Molecules transformed by  $2_1$  symmetry form an infinite chain along the *a* axis owing to a branched H-bond (O1'...H-O2 and O2'...H-O2). The parameters of this bond are:  $2.90$ ,  $2.22$ ,  $161.3$  and  $2.92$ ,  $2.40$ ,  $131.6$ .

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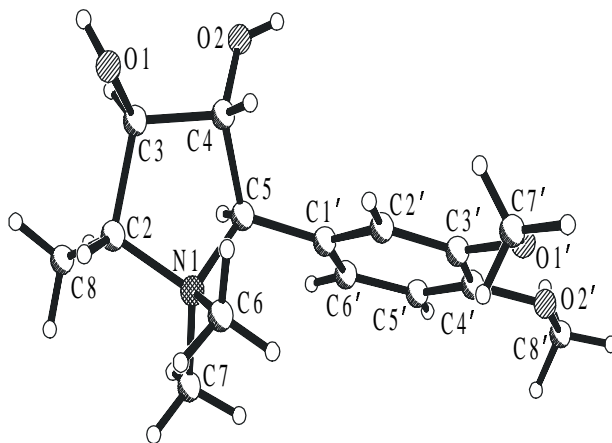


Fig. 1. Conformation and atomic numbering in **3**.

In conclusion, it should be noted that **2** has a N-containing ring similar to codonopsine (according to PMR spectroscopy they are stereochemically identical) and differs only by the presence of a second OCH<sub>3</sub> group on the aromatic ring. Therefore, the absolute configuration of codonopsine probably corresponds with that of codonopsinine.

Thus, the alkaloids codonopsine and codonopsinine have structures and absolute configurations N-methyl-2(*R*)- $\alpha$ -methyl-3(*R*)- $\beta$ -hydroxy-4(*R*)- $\alpha$ -hydroxy-5(*R*)- $\beta$ -(3,4-dimethoxyphenyl)pyrrolidine and N-methyl-2(*R*)- $\alpha$ -methyl-3(*R*)- $\beta$ -hydroxy-4(*R*)- $\alpha$ -hydroxy-5(*R*)- $\beta$ -(4-methoxyphenyl)pyrrolidine.

Single crystals of codonopsine methyl iodide (C<sub>15</sub>H<sub>24</sub>NO<sub>4</sub>I, mp 207-208 °C) were obtained from methanol. The crystals were transparent elongated prisms. Unit-cell constants and intensities of reflections were measured on a STOE Stadi-4 four-circle diffractometer ( $\theta/2\theta$ -scanning) using Mo K $\alpha$ -radiation (graphite monochromator):  $a = 7.805(2)$ ,  $b = 13.592(3)$ ,  $c = 16.667(3)$  Å,  $V = 1768.1(6)$  Å<sup>3</sup>,  $\rho = 1.537$  g/cm<sup>3</sup>, space group  $P2_12_12_1$ ,  $Z = 4$ . Absorption corrections were not applied.

The structure of **3** was solved by direct methods using the SHELXS-97 programs. The structure was refined using the SHELXL-97 programs. Nonhydrogen atoms were refined by anisotropic full-matrix least-squares methods (over  $F^2$ ). The positions of H atoms were found geometrically and refined isotropically with fixed thermal parameters  $U_{\text{iso}} = nU_{\text{eq}}$ , where  $n = 1.5$  for methyls and 1.2 for others and  $U_{\text{eq}}$  is the equivalent isotropic thermal parameter of the corresponding C atoms. H-atoms of hydroxyls were found from a difference electron-density synthesis and refined isotropically. The final agreement factor  $R_1 = 0.0369$ ,  $wR_2 = 0.0798$  for 2164 reflections with  $I > 2\sigma(I)$  and  $R_1 = 0.0416$  and  $wR_2 = 0.0862$  for the whole data set (2439).

The data from the XSA were deposited as a CIF file in the Cambridge Structural Database (CCDC 249723).

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